

NEW ANTIMALARIAL AND ANTITUBERCULOSIS DRUGS

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NEW ANTIMALARIAL AND ANTITUBERCULOSIS DRUGS

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I. ABSTRACT

The non-mevalonate biosynthetic pathway has only recently been mapped and identified as a potential target for the development of novel antibiotics. With the current problems in regard to antibiotic resistance, the discovery of new compounds that are active medicinally against bacterial infections like malaria and tuberculosis, that cause the death of approximately two million people worldwide yearly, is imperative. The goal of this project was aimed at the synthesis and evaluation of methylerythritol cyclodiphosphate (MEcPP) and 4-hydroxy-3-methyl-2-(E)-butenyl-4-diphosphate (HMBPP) analogs as potential therapeutic agents against bacterial infections such as malaria, tuberculosis, and related infections. In bacteria and plants, the non-mevalonate biosynthetic pathway is utilized to produce crucial isoprenoid building blocks. This pathway is appealing as a target for the development of new antibiotics because it is absent in humans, likely resulting in reduced toxicity and side effects. Particularly, MEcPP analogs may be especially selective due to the unique 8-membered cyclic pyrophosphate moiety in MEcPP. An MEcPP analog has never been synthesized before, and this project intended to prepare and evaluate both cyclic and acyclic MEcPP and HMBPP mimics. The goal of this project was to take steps toward filling the gaps in current knowledge that may allow for the production of antibiotics, specifically those of medicinal relevance with activity on resistant bacterial strains.

II. ACKNOWLEDGMENTS

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TABLE OF CONTENTS

List of Figures and Schemes.....	6
List of Abbreviations.....	7
Introduction.....	8
Results and Discussion.....	12
Conclusion and Future Work.....	15
Experimental Section.....	17
References.....	21

LIST OF FIGURES

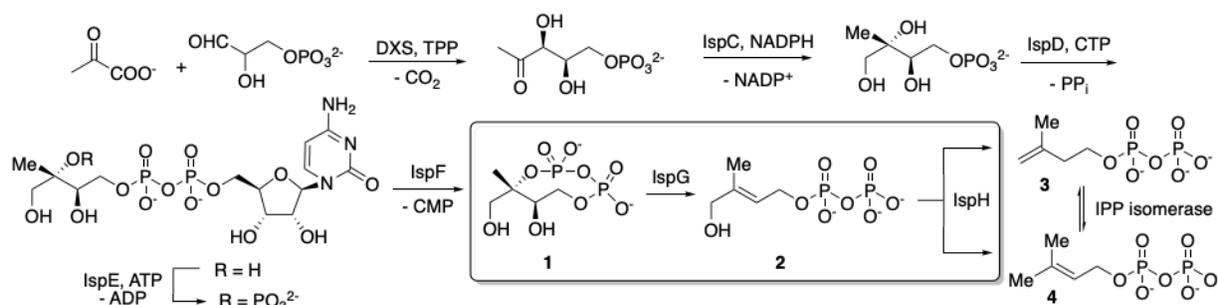
Scheme 1. Non-mevalonate pathway.....	8
Figure 1. Various potential MEcPP analogs.....	9
Figure 2. Various potential HMBPP analogs.....	10
Scheme 2. Enzyme inhibition.....	11
Scheme 3. MEcPP and HMBPP mimics.....	12
Scheme 4. First synthetic step: Acetal formation.....	13
Scheme 5. Second synthetic step: Hydrolysis of the alcohol to an aldehyde	13
Scheme 6. The Grignard Reaction.....	14
Scheme 7. Third synthetic step: Alkylation via the Grignard Reaction.....	15
Figure 3. Additional potential target molecule.....	16
Scheme 8. Overall synthesis of the target MEcPP analog.....	16
Figure 4. ^1H NMR of the product from the first step.....	18
Figure 5. ^1H NMR of the product from the second step.....	19
Figure 6. ^1H NMR of the product from the third step.....	20

LIST OF ABBREVIATIONS

¹H NMR	<i>Proton Nuclear Magnetic Resonance</i>
DCM	<i>Dichloromethane</i>
DMF	<i>Dimethylformamide</i>
Eq or Equiv	<i>Equivalent</i>
Et₃B	<i>Triethylborane</i>
EtOAc	<i>Ethyl acetate</i>
H₃O⁺	<i>Hydronium (acid)</i>
HMBPP	<i>4-hydroxy-3-methyl-2-(E)-butenyl-4-diphosphate</i>
Hrs	<i>Hours</i>
K₂CO₃	<i>Potassium bicarbonate</i>
KMnO₄	<i>Potassium permanganate</i>
Me₄Si	<i>Tetramethylsilane</i>
MEcPP	<i>Methylerythritol cyclodiphosphate</i>
MeOH	<i>Methanol</i>
MEP	<i>Non-mevalonate pathway</i>
MgSO₄	<i>Magnesium sulfate</i>
N₂	<i>Under Nitrogen gas</i>
NaH₂PO₂	<i>Sodium hypophosphite</i>
NaOH	<i>Sodium hydroxide</i>
NCS	<i>N-chlorosuccinimide</i>
NH₄Cl	<i>Ammonium chloride</i>
PCC	<i>Pyridinium chlorochromate</i>
R	<i>Any group</i>
R.t.	<i>Room temperature</i>
TEMPO	<i>(2,2,6,6-Tetramethylpiperidin-1-yl)oxyl or (2,2,6,6-Tetramethylpiperidin-1-yl)oxidanyl</i>
THF	<i>Tetrahydrofuran</i>
TLC	<i>Thin layer chromatography</i>

III. INTRODUCTION

Biosynthesis of isoprenoid precursors via the non-mevalonate pathway is opportunistic in designing and synthesizing novel antibiotics and antiparasitic agents for medical use, specifically in the cases of tuberculosis and malaria. The non-mevalonate (methylerythriol-4-phosphate or MEP) (Scheme 1) pathway was only recently discovered. This pathway is compelling because it



Scheme 1. The non-mevalonate pathway for isoprenoid precursor biosynthesis (methylerythritol cyclodiphosphate (MEcPP) **1**, 4-hydroxy-3-methyl-2-(*E*)-butenyl-4-diphosphate (HMBPP) **2**, isopentenyl diphosphate (IPP) **3**, dimethylallyl diphosphate (DMAPP) **4**).

does not exist in humans – allowing for selective inhibition in the pathogen.¹ The isoprenoids that ultimately result from this pathway are converted into critical lipid metabolites and therefore are crucial for cell growth.² The MEP pathway is used by bacteria, parasites, and plants, which indicates its utility in tuberculosis and malaria cases. The World Health Organization (WHO) estimates that in 2019, 10 million people worldwide had tuberculosis, resulting in 1.5 million deaths. TB is one of the top ten causes of death, and the leading cause of death from a single infectious agent – even higher than HIV/AIDS. In regard to malaria, the WHO estimates 228 million cases in 2019, resulting in 405,000 deaths.³ The WHO further states that "multidrug-resistant TB (MDR-TB) remains a public health crisis and a health security threat". TB is also the leading killer of HIV-infected people. Due to increased resistance in a variety of pathogens including the aforementioned, drugs must continuously be developed to target them effectively.

The MEP pathway is a promising source of opportunity in novel drug development. There have been very few inhibitors of the non-mevalonate pathway reported thus far, and those that have been were reviewed in 2013.^{4,5} According to a literature survey, there is a scarcity of inhibitors for the MEP pathway.

The goal of this research was to synthesize analogs of the metabolites MEcPP (1) (Figure 1) and HMBPP (2) (Figure 2), the substrates of the IspG and IspH enzymes respectively, in hopes to achieve novel bioactivity. The specific goal analog is boxed in Figure 1.

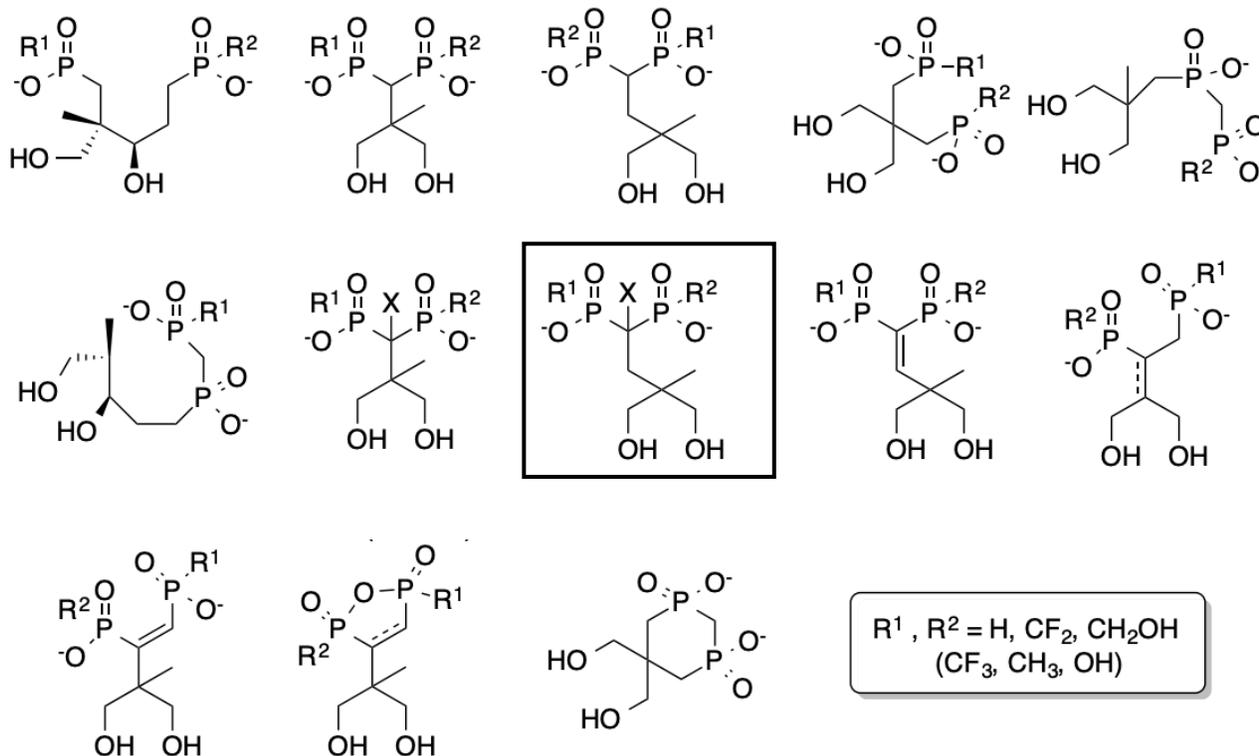


Figure 1. Various potential MEcPP analogs.

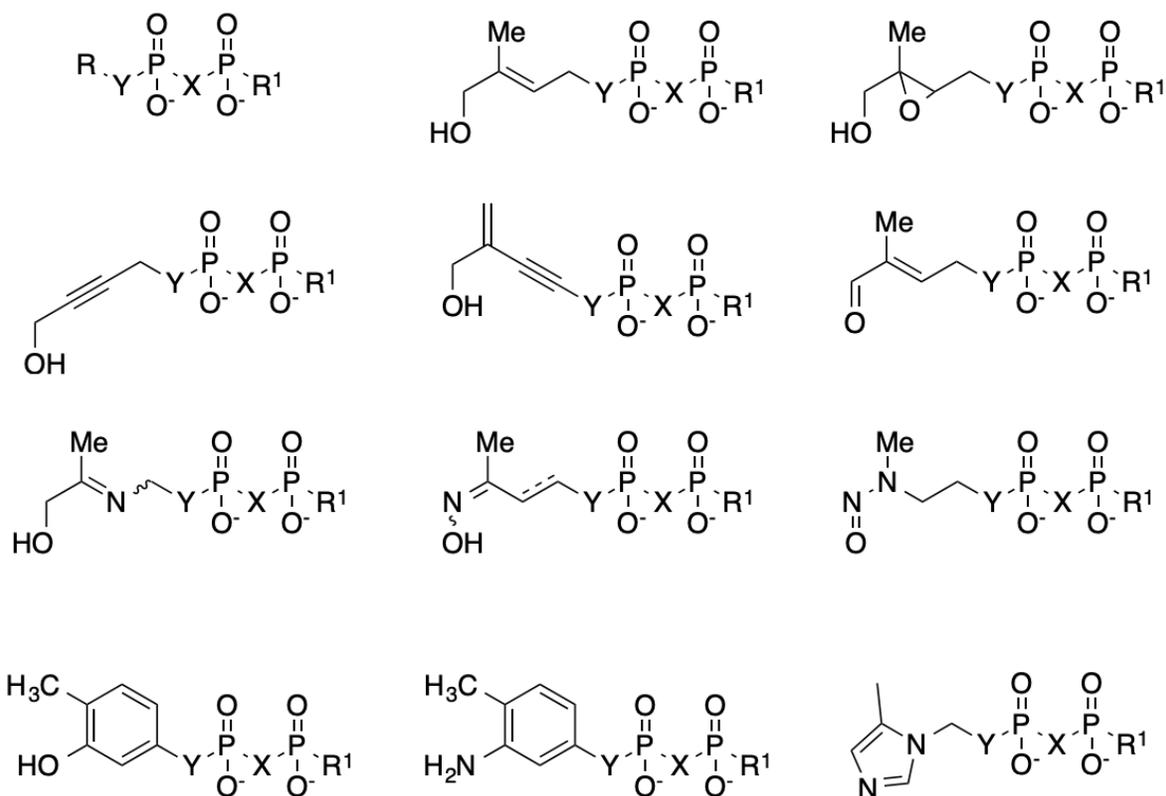
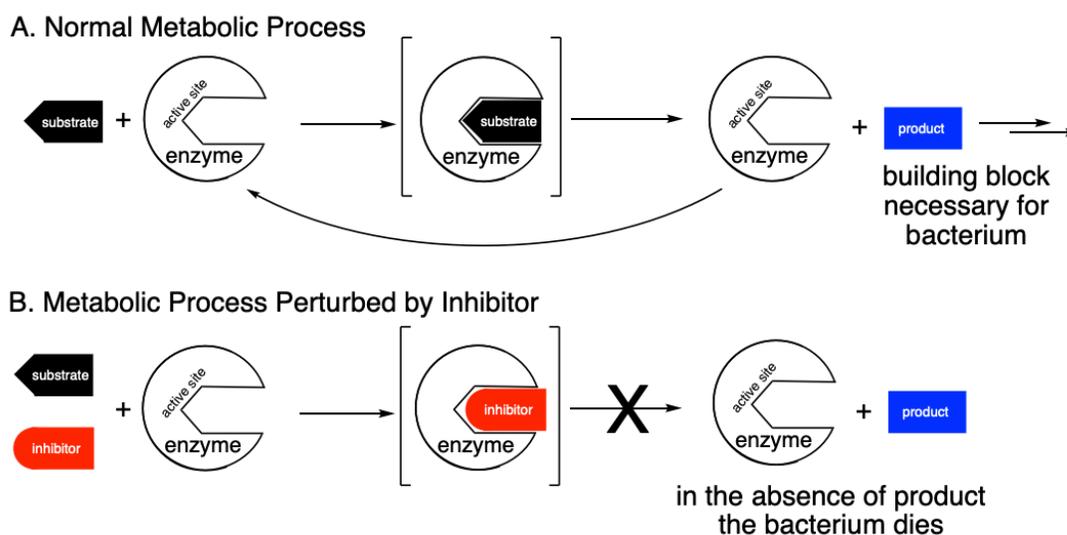


Figure 2. Various potential HMBPP analogs.

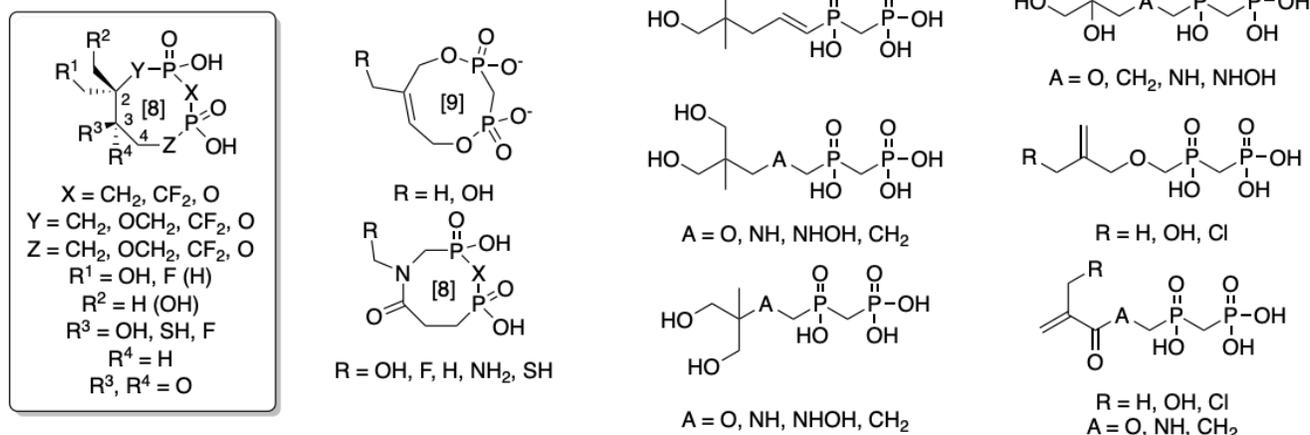
Blocking the machinery that creates these substrates will lead to bacterial death because of their necessary role in bacterial growth. No cyclic pyrophosphate analog of MEcPP has ever been prepared for the inhibition of the IspG enzyme. Similarly, there have not been any analogs of acyclic HMBPP tested to inhibit the enzyme IspH. The more similar the synthesized compound to the natural metabolite, the higher the likelihood of strong inhibition of the enzymes involved. The goal is to make the inhibition strong enough to restrict the organism's (bacteria or parasite) ability to produce key molecules necessary for survival. This will result in death of the organism. The inhibitor interacts with the enzyme in a way that prevents it from carrying out its function – processing its normal substrate. Therefore, there will not be enough product made to support the



Scheme 2. Blockade of a metabolic pathway with an inhibitor that looks like the substrate of an enzyme. The enzyme cannot perform its normal function or be recycled, therefore the resulting product is depleted ultimately leading to the organism's death.

organism's survival (Scheme 2). This mode of action – enzymatic inhibition – is the most common among all drugs. 47% of all current drugs inhibit enzymes.⁶ Unlike many other drugs' enzymatic targets, humans do not possess the target (IspG and IspH). As a result, the compounds are much less likely to interfere with human pathways. This is beneficial in that possible toxicity or side-effects should be minimized.

The synthesis of a variety of compounds mimicking the overall structure of MEcPP and HMBPP is the ultimate goal of this ongoing research. Scheme 3 displays a few of the possible target structures. Because the cyclic targets are significantly more difficult to make, the focus was on the acyclic targets. Once synthesized, the compounds can be sent for testing against IspG and IspH in Professor Andrew Wiemer's laboratory at the University of Connecticut.



Scheme 3. Representative target molecules, analogs of MEcPP and HMBPP.

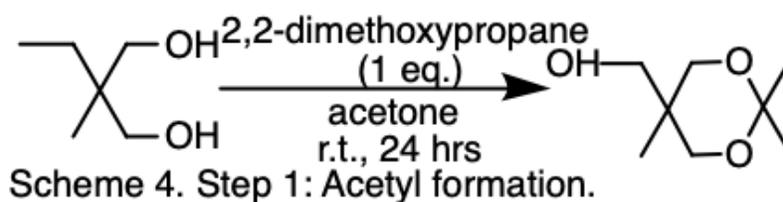
IV. RESULTS AND DISCUSSION

Although the target HMBPP analog was not obtained, progress was made in its synthesis.

The first steps were accomplished in synthesizing this target molecule and subsequent steps were planned.

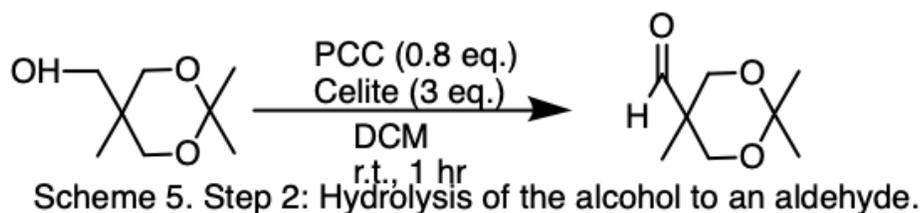
1. Acetal Formation

The first step of the synthesis involved a reaction between tris(hydroxymethyl)ethane and acetone/2,2-dimethoxypropane as seen in Scheme 4. The purpose of this step was a protection step. The ring formation limited the molecule to only one hydroxyl group available for reaction. This step was first carried out with only acetone initially and was unsuccessful. But, following the addition of the 2,2-dimethoxypropane to the procedure, this reaction resulted in 95 percent yield. The product was not purified because based on the ^1H NMR, the purity was sufficient to utilize in the next step.



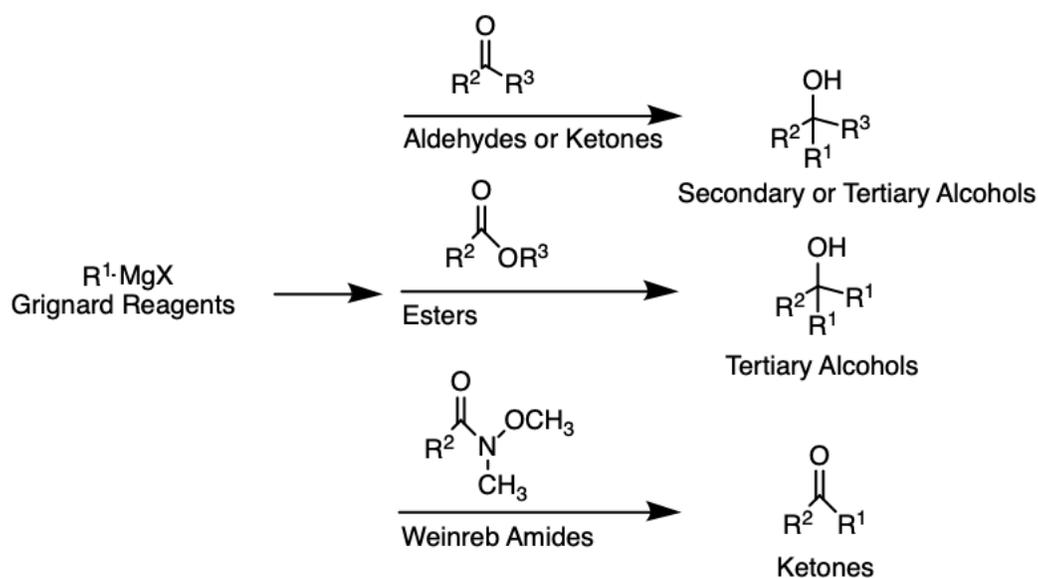
2. Hydrolysis of the Alcohol to an Aldehyde

The next step in the synthesis involved oxidation of the hydroxyl group to yield the aldehyde, as shown in Scheme 5. The formation of the aldehyde is an important step because it acts as an intermediate for many reactions including the following step in this synthesis, alkylation. Scheme 5 shows the conditions that were utilized in the oxidation. This step was particularly difficult. Multiple methods of synthesis were attempted previous to the PCC/celite method. The different syntheses attempted included one utilizing N-chlorosuccinimide, dodecyl methyl sulfide, and triethylamine and one that included sodium hypochlorite and TEMPO. Both resulted in the aldehyde, but with low yields between 5 and 40 percent. Finally, the oxidation using a PCC/celite mixture in dichloromethane proved to be a successful synthesis. This reaction resulted in an 81 percent yield. Because the aldehyde is an intermediate in this sequence of reactions, and the ^1H NMR showed the product to be relatively clean, it was not purified and used directly in the next step.



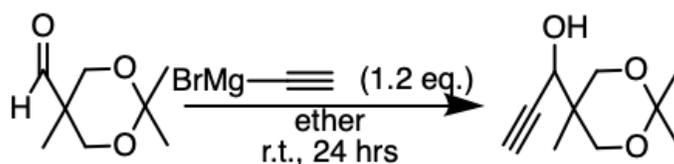
3. Alkylation

Following the aldehyde formation, an alkylation reaction was carried out to introduce another carbon-carbon bond into the molecule. Although a method utilizing lithium acetylide-ethylenediamine complex was attempted, the Grignard reaction proved to give a better yield. This reaction is used to introduce alkyl groups into molecules. It can be utilized with functional groups such as aldehydes and ketones (as in this reaction), esters, and Weinreb amides to yield secondary or tertiary alcohols, tertiary alcohols, and ketones respectively, as shown in Scheme 6.⁷ The



Scheme 6. The Grignard Reaction

specific Grignard reagent shown in Scheme 7 was used due to the resulting product's utility in the next step of the synthesis discussed in the 'Future Work' section. This reaction resulted in a 70 percent yield.



Scheme 7. Step 2: Alkylation via the Grignard reaction.

The ^1H NMR spectra and data for the three products of each of the synthesis steps are found in the ‘Experimental’ section. Unfortunately, the remaining steps in this synthesis were not carried out due to Covid-19 and TCU’s transition to online learning, but the planned remaining steps are found in the ‘Conclusion and Future Work’ section.

V. CONCLUSION AND FUTURE WORK

The goal of the synthesis of the target MEcPP analog did not occur, but the first three steps showed promise in achieving its synthesis in the future. The yields of the initial three steps were relatively high, which will be advantageous in continuing the synthetic scheme. Scheme 8 shows the overall synthesis of the target analog including both the first three steps that were successfully completed and the future experimental steps to be carried out.

The urgent need for antibiotics and antiparasitic agents against multidrug-resistant pathogens underlines the necessity to continue the work on this project. The non-mevalonate pathway for the biosynthesis of isoprenoid precursors (Scheme 1) present in bacteria, parasites, and plants offers opportunities for the synthesis of these medications against major diseases like tuberculosis and malaria. Further research should continue to focus on the synthesis not only on the analog that was the primary subject of this project, as well as another potential target via Wadsworth-Horner-Emmons olefination of the aldehyde (product in Scheme 5), shown in Figure

3; but also on other possible inhibitors – additional analogs of MEcPP, the substrate of the enzyme IspG and analogs of HMBPP, the substrate of the enzyme IspH, both of which play integral roles in the MEP pathway.

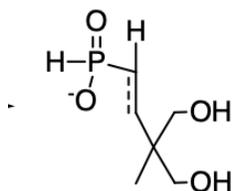
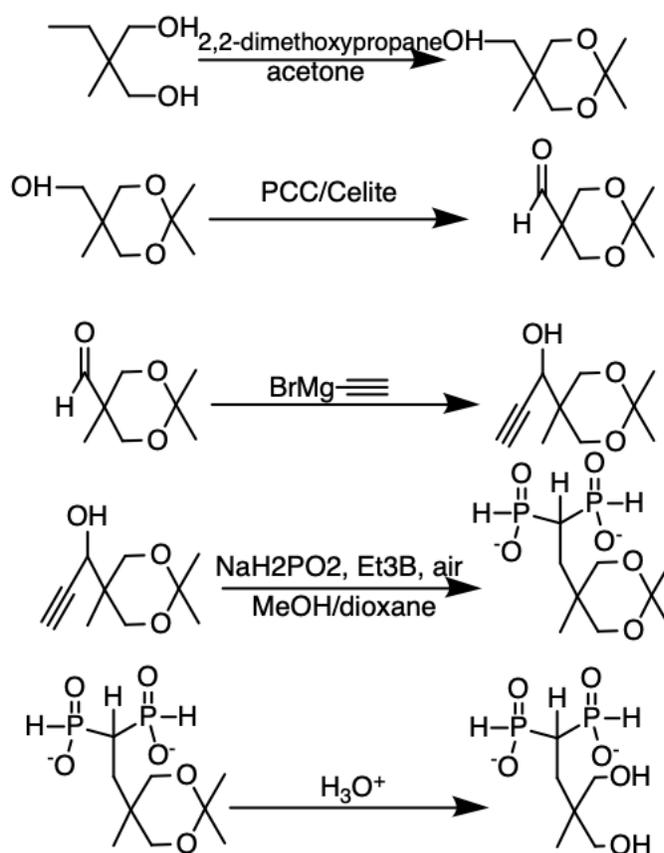


Figure 3. Additional potential target molecule through the Wadsworth-Horner-Emmons olefination of the aldehyde.



Scheme 8. Overall synthesis of the target MEcPP analog.

VI. EXPERIMENTAL SECTION

General Chemistry:

¹H NMR spectra were recorded on a 300-MHz Varian INOVA spectrometer or 400-MHz Bruker Avance spectrometer. ¹H NMR were acquired using CDCl₃, and chemical shifts are reported in part per million relative to internal Tetramethylsilane (Me₄Si, δ = 0.00 ppm). Flash chromatography was carried out on Silica Gel premium Rf grade (40–75 μ m). Ethyl acetate/hexane mixtures or dichloromethane/acetone were used as the eluent for chromatographic purifications. TLC plates were visualized by UV or immersion in potassium permanganate solution (3 g KMnO₄, 20 g K₂CO₃, 5 mL 5% aq. NaOH and 300 mL of water) followed by heating.

Reagents and solvents:

All starting materials were purchased from commercial sources and used as received. The solvents were distilled under N₂ and dried according to standard procedures (DMF from MgSO₄; CH₃CN, toluene and dichloromethane from CaH₂).

First Step – Acetal Formation⁸

A 1000 mL round bottom flask containing acetone (64 equiv, 250 mL, 3404 mmol) was flushed with N₂. Next, 2,2-dimethoxypropane (1 equiv, 30 mL, 250 mmol), tris(hydroxymethyl)ethane (1 equiv, 30 g, 250 mmol), and p-Toluenesulfonic acid (0.01 equiv, 0.48 g, 2.5 mmol) were added to the round bottom flask. The mixture was stirred for 24 hours in room temperature under N₂. Then, the reaction was quenched with K₂CO₃ (0.35 equiv, 12.0 g, 87 mmol) and the solvent was evaporated. Figure 4 shows the ¹H NMR of this product. The NMR indicated that the product was the intended molecule based on the peak data shown below, and because the spectrum is relatively clean, the product was pure enough to proceed with the next step.

^1H NMR (400 MHz, Chloroform- d) δ 3.66 (d, $J = 7.8$ Hz, 2H), 3.64 – 3.63 (m, 2H), 3.58 (d, $J = 12.0$ Hz, 2H), 2.85 – 2.74 (m, 1H), 1.42 (s, 3H), 1.37 (s, 3H), 0.81 (s, 3H).

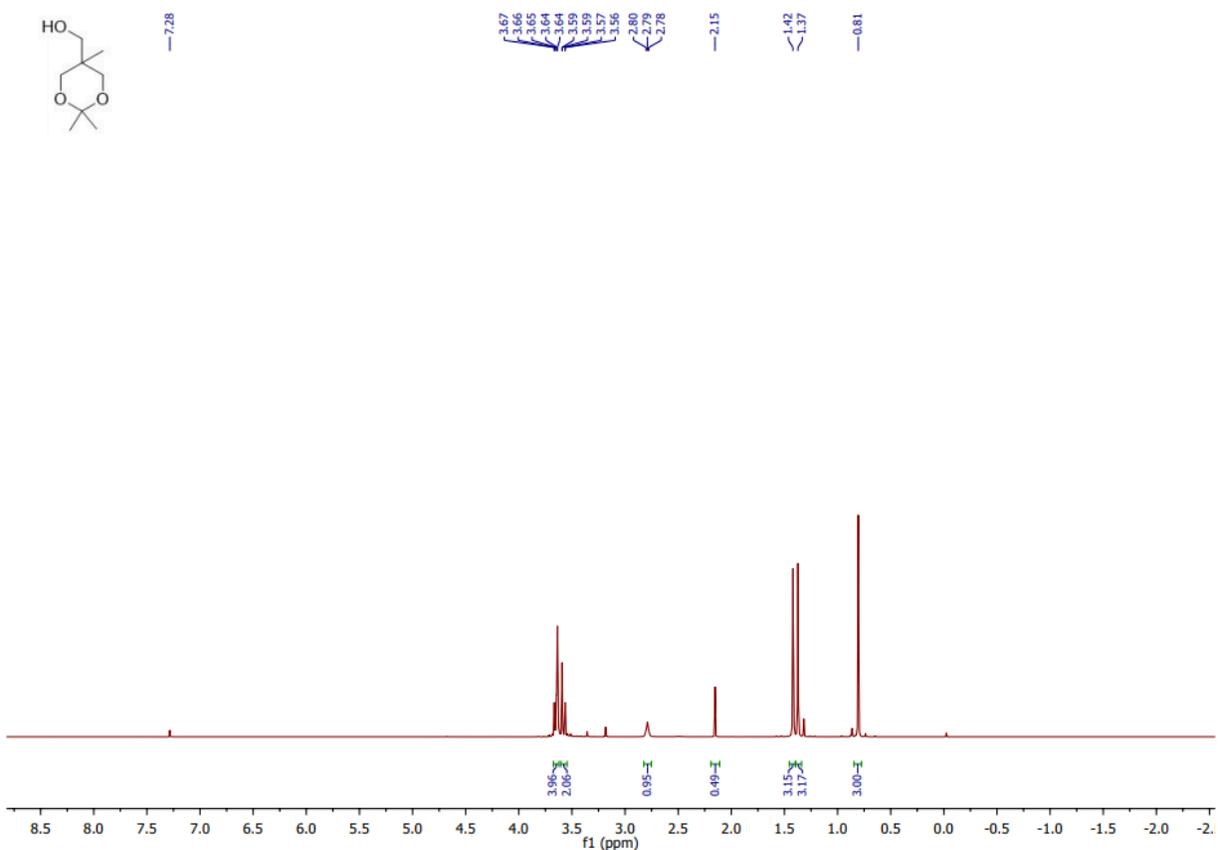


Figure 4. ^1H NMR of the product from the first step.

Second Step – Hydrolysis of the Alcohol to an Aldehyde

350 mL of DCM was added to a 1000 mL round bottom flask and flushed with N_2 . The product from the first step (1 equiv, 13.8 g, 84 mmol) was then added and dissolved. The 50/50 by mass mixture of PCC (0.8 equiv, 15.1 g, 70 mmol) and celite (3 equiv, 15.1 g, 250 mmol) was slowly added to the round bottom flask. The mixture was stirred for 1 hour in room temperature under N_2 . Then, the mixture was filtered to remove the precipitate and the solvent was evaporated. Figure 5 shows the ^1H NMR of this product. The NMR indicated that the product was the intended molecule

based on the peak data shown below, and because the spectrum is relatively clean, the product was pure enough to proceed with the next step.

^1H NMR (400 MHz, Chloroform- d) δ 9.74 (s, 1H), 4.04 (m, $J = 11.9, 0.9$ Hz, 2H), 3.81 – 3.63 (m, 2H), 1.41 (d, $J = 0.7$ Hz, 3H), 1.30 (d, $J = 0.8$ Hz, 3H), 0.82 (s, 3H).

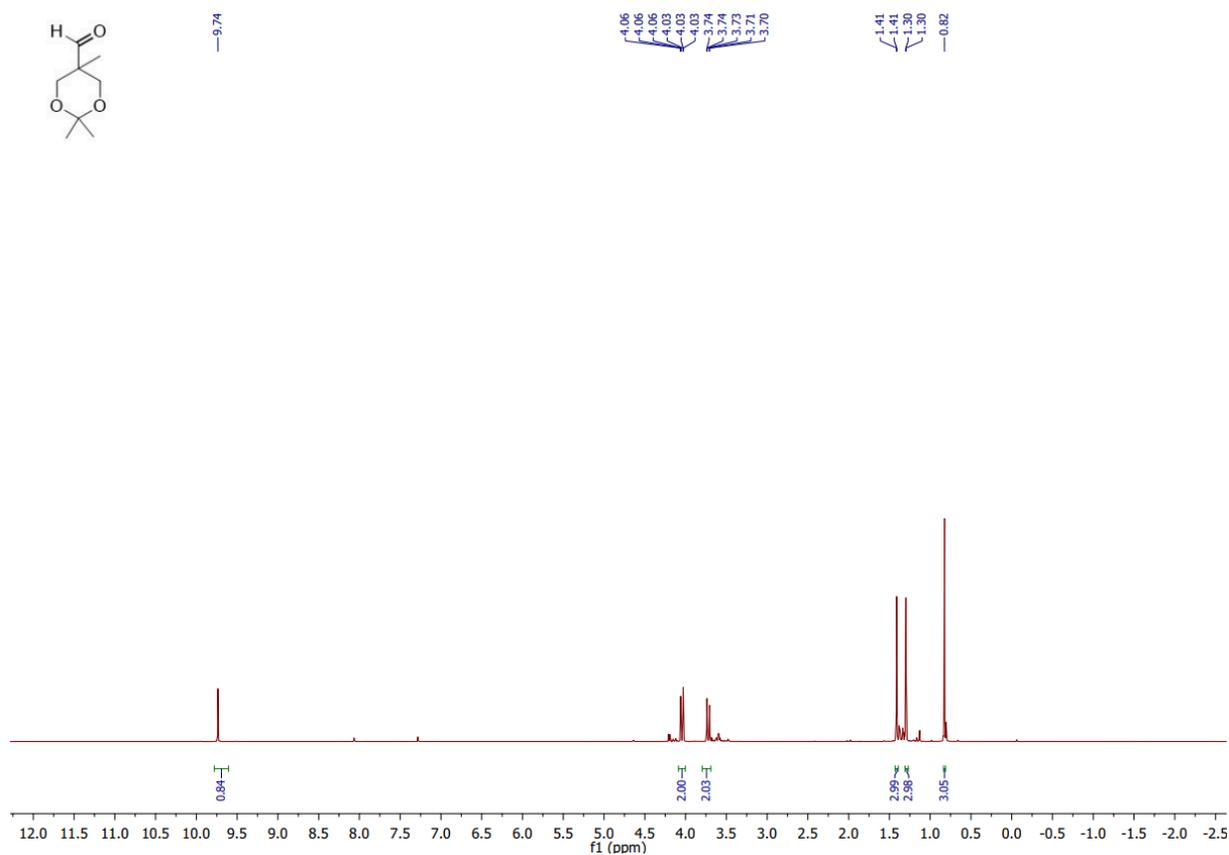


Figure 5. ^1H NMR of the product from the second step.

Third Step – Alkylation

50 mL of THF was added to a 500 mL round bottom flask and flushed with N_2 . The product from the second step (1 equiv, 1 g, 6 mmol) was then added and dissolved. Next, the ethynylmagnesium bromide (0.5M in THF) (1.2 equiv, 14 mL, 7 mmol) was added slowly. The mixture was stirred

for 24 hours in room temperature under N_2 . Then, the reaction was quenched with NH_4Cl and washed with ethyl acetate and brine three times. The organic layer was dried with $MgSO_4$, filtered, and evaporated to remove the solvent. Figure 6 shows the 1H NMR of this product.

No data available.

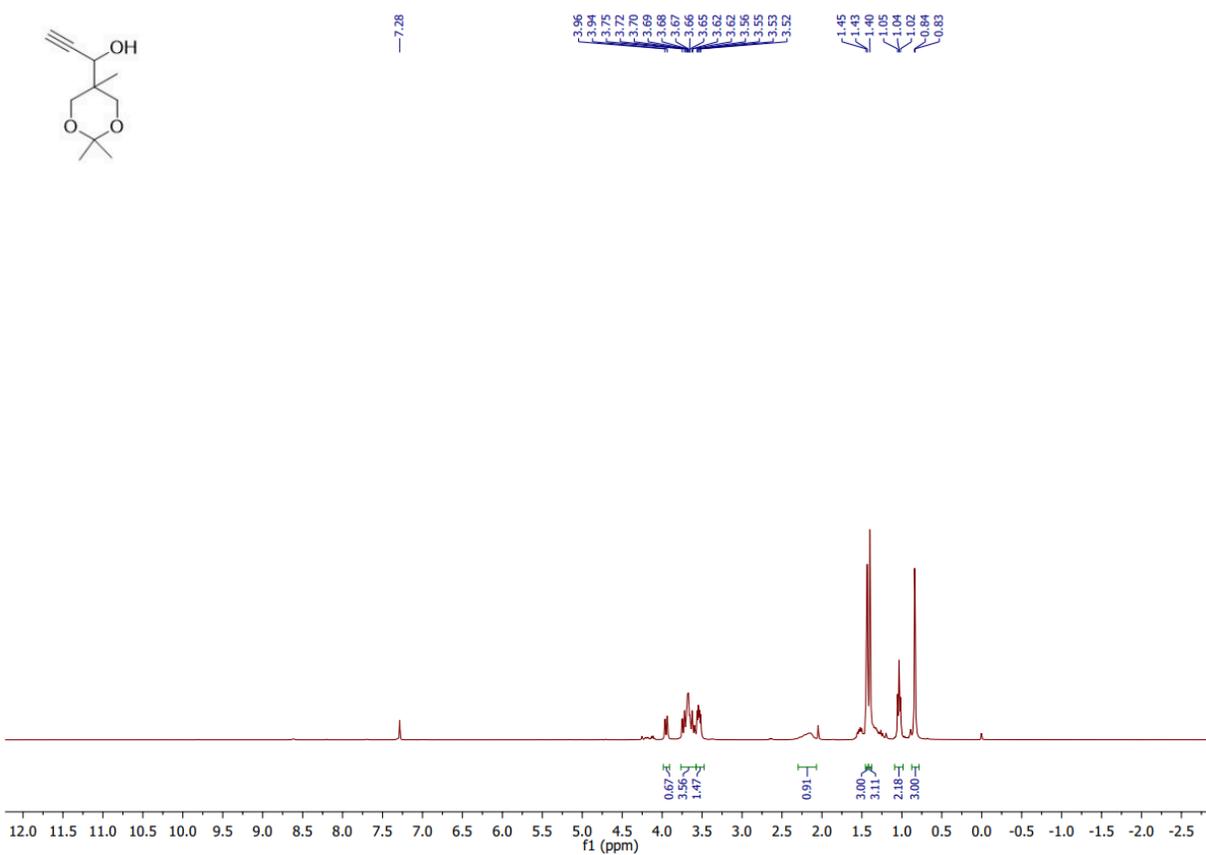


Figure 6. 1H NMR of the product from the third step.

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